# NEW FURANOEREMOPHILANES FROM SOUTH AFRICAN SENECIO SPECIES\*

FERDINAND BOHLMANN and CHRISTA ZDERO
Institute of Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, West Germany

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Abstract—An investigation of South African Senecio species afforded in addition to previously known compounds three new furanoeremophilanes, six compounds derived from cacalol and a dihydroeuparin derivative. Structures were elucidated by spectroscopic methods and some chemical transformations. Chemotaxonomical aspects are discussed briefly.

#### INTRODUCTION

During the last three years we have investigated several representatives of the large genus Senecio [1, 2]. As a continuation of our studies of South African species we have now investigated some additional plants, all collected in Natal. The results confirm that furanoere-mophilanes are very widespread in this genus.

#### RESULTS AND DISCUSSION

Senecio medley-woodii Hutch. was collected near St. Michels-on-Sea in Natal. The roots contained as the main constituent  $10\beta$ -furanoeremophilane (1) [3] together with two further compounds, which have not been isolated before. The <sup>1</sup>H NMR data of the less

polar compound showed it was a diester of 1. A singlet at  $\delta$  2.04 (3H), a doublet at 1.22 (3H) (J = 7 Hz), a tq at 2.42 (2H) (J = 7 and 7) and a triplet at 0.94 (3H) (J = 7)clearly indicated the nature of the ester groups. Signals at 5.27 (1H, ddd, J = 9, 4, 4) and at 6.35 (br s) together with those typical for a furanoeremophilane (see Table 1) showed, that the ester groups were located at C-3 and C-6. Partial saponification afforded a compound with a  $3\beta$ -hydroxyl and a  $6\beta$ -(2-methylbutyric acid) ester. Therefore the diester had structure 2. The cis-annelation followed from the observed coupling constants (see Table 1). The third compound was identified as the corresponding ketone 4. As observed earlier [1, 2], it is only at elevated temperature that compounds of this type afford first order NMR spectra while at room temperature broad signals are observed. Partial saponification showed that the ester groups were in the same position as in 2 (Table 1). The aerial parts contained germacrene D (6) in addition to 1 and 2.

The roots of S. halimifolius L., one of the succulent

Table 1. <sup>1</sup>H NMR signals of compounds 2-5 (270 MHz, δ-values, TMS as internal standard)

	2 (CDCl <sub>3</sub> )	3 (CDCl <sub>3</sub> )	$4 (C_6 D_6, 75^\circ)$	$5 (C_6 D_6, 75^\circ)$
3α-H	5.27 ddd	4.35 ddd	5.00 ddd	3.63 m
4α-H	$2.10 \ m$	2.04 m	$1.97 \ m$	$2.00 \ m$
6α-H	$6.35 \ s(br)$	$6.36 \ s(br)$	$6.37 \ s(br)$	$6.39 \ s(br)$
9α- <b>H</b>	$2.32 \ d(br)$	$2.26 \ d(br)$	-	_
9β-H	2.84 dd(br)	2.87 dd(br)	_	
10β-H	1.88 m	1.8 m	2.63 dd	2.62 dd
12-H	$7.04 \ s(br)$	$7.05 \ s(br)$	$6.85 \ s(br)$	$6.84 \ s(br)$
13-H	1.87 d	1.86 d	$1.81 \ s(br)$	1 82 s(br)
14-H	1.05 s	1.06 s	1.16 s	1 21 s
15-H	0.95 d	1.01 d	0.76 d	0.81 d
OAc	2.04 s	_	1.68 s	<del>-</del>
OCOR	2.42 ta	2.45 ta	2.26 tg	2.26 tq
	1.22 d	$1.24 \ d$	$1.08 \ d$	1.05 d
	0.94 t	1.02 t	$0.78 \ t$	$0.80 \ t$

J (Hz); 2 and 3:  $2\alpha,3\alpha = 3\alpha,4\alpha = 4$ ;  $2\beta,3\alpha = 9$ ;  $4\alpha,15 = 7$ ;  $9\alpha,9\beta = 17$ ;  $9\beta,10\beta = 6$ ; 12,13 = 1; 4 and 5:  $1\alpha,10\beta = 7$ ,  $1\beta,10\beta = 4$ ;  $4\alpha,15 = 7$ 

<sup>\*</sup>Part 143 in the series "Naturally Occurring Terpene Derivatives", for part 142 see: Bohlmann, F., Mahanta, P. K., Jakupovic, J., Rastogi, R. C. and Natu, A. A. (1978) Phytochemistry 17, 1165.

species contained in very minute amounts a furanoere-mophilane, whose spectroscopic properties were very similar to those of 7 [4]. However the 6-H signal was shifted to considerably higher field. The only structure which agrees with this observation is  $\bf 8$ , the  $\bf 6\alpha$ -epimer of 7 (Table 2).

The aerial parts afforded 6, 9, 10 [5] and 13 [6] together with a further euparin derivative, with structure 11, as shown by saponification of 10, which yielded an alcohol identical with 11.

Senecio inaequidens DC has been investigated before

[1, 2]. However, the material was grown from seeds. Reinvestigation of material collected from several places in Natal showed that there were some differences in the composition of the constituents. Most of the compounds isolated previously were found again. However, none of the acyl pyrroles reported before could be isolated. The aerial parts yielded in addition to the typical benzofurans 13, 14, 16 and 23 [1, 2, 6] three new ones. The structure of 15 was elucidated from the NMR spectral data, which were very similar to those of 14 (Table 2). The two other compounds could be

	8	+ Eu(fod) <sub>3</sub>	15	18	19	20
1α-Η 1β-Η			6.94 dd	2.83 dd(br) 2.54 m	2.85 dd(br) 2.60 m	2.84 dd(br) 2.60 m
2-Н	2.64 m	7.75 m	5.94 ddd )		)	)
3-H		$\begin{cases} 3.72 \ m \\ 3.14 \ m \end{cases}$	$ \begin{array}{c} 2.54 \ dddd \\ 2.15 \ m \end{array} $	1.80 m	$\begin{cases} 1.81 \ m \end{cases}$	\\ 1.81 m
4-H	2.29 ddg	4.33 m	3.33 dq(br)	3.25 ddq	3.41 ddq	3.46 ddg
6-H	5.93 s	$7.13 \ s$				_
9- <b>H</b>	7.44 s	12.28 s		_		
12-H	$7.26 \ q$	$7.40 \ s(br)$	7.37 a	7.22  q	7.30 a	$7.29 \ q$
13-H	$2.10 \ \hat{d}$	$2.47 \ s(br)$	$2.33 \hat{d}$	2.37 d	2.29 d	2.41 d
14-H	1.02 s	2.47 s(br)	{4.45 d }4.36 d	2.57 s	§ 5.49 d § 5.37 d	5.01 d 4.95 d
15-H	1.03 d	1.88 d	1.12 d	1.19 d	1.23 d	1.27 d
OCH,		_	4.13 s		_	
OCOR	6.01 aa	6.25 qq	$2.27 \ d(br)$	2.70 a	2.71 q	2.71 a
	1.85 dq	2.32 d(br)	2.15 m	1.33 t	1.34 t	1.34 t
	1.80 dq	2.23 s(br)	0.96 d	_		
OAc	_		<del>-</del>		2.09 s	_

Table 2. <sup>1</sup>H NMR data for compounds 8, 15, 18, 19 and 20 (CDCl<sub>2</sub>)

J (Hz); **8**: 3 $\alpha$ ,4 $\alpha$  = 6; 3 $\beta$ ,4 $\alpha$  = 9; 4 $\alpha$ ,15 = 7; 12,13 = 1; 18,19 = 7; 18,20 = 19,20 = 1.5. 15: 1.2 = 9; 1.3 = 3; 2.3 = 6.5; 2.3' = 2.3; 3.4 = 7; 4.15 = 7; 12,13 = 1.4; 14,14 = 13; 17,18 = 8; 18,19 = 18,20 = 7; 18-20: 1 $\alpha$ ,1 $\beta$  = 16; 1 $\alpha$ ,2 $\alpha$  = 4; 3 $\alpha$ ,4 $\alpha$  = 5; 3 $\beta$ ,4 $\alpha$  = 10; 4 $\alpha$ ,15 = 7. 19 and 20: 14,14 = 13.

separated from 23 only with difficulty. The structure of 24 followed from the NMR spectral data and the data of the anhydro compound 28, obtained by heating 24 with acetic anhydride. Compound 25, however, could not be separated completely from 23. After acetylation the NMR spectra (Table 3) of the resulting acetates showed that we were dealing with the structures 26 and 27. As demonstrated earlier, ketones like 23 are acetylated via enolisation [1, 2]. Therefore the presence of 25 seems to be established.

The roots of Senecio inornatus DC afforded large quantities of 13, cacalol (17) [7] and 21 together with two further benzofurans (18 and 19). The structures clearly followed from the spectroscopic data. The relative position of the acetate group was indicated by the chemical shift of the corresponding NMR signal (Table 2). The aerial parts also contained 17, 18 and 19 together with 6 and a further benzofuran, which was found to be 20 since acetylation produced compound 19.

The chemotaxonomical importance of the compounds isolated from the four species is not completely clear. Furanoeremophilanes are definitely typical for the

Table 3. <sup>1</sup>H NMR data for compounds 24, 26, 27 and 28 (CDCl<sub>3</sub>)

	24	$\Delta^+$	26	27	28
2-H	6·70 s	0.14	7.04 s	7.15 s	7.02 s
12-H	7.33 q	0	7.36 q	7.45 q	7.46 q
13-H	$2.29 \ \hat{d}$	0	2.44 d	$2.39 \ \bar{d}$	$2.33 \ s(br)$
14- <b>H</b>	$2.73 \ s$	0.02	3.09 s	5.79 s	$2.70 \ s$
15-H	1.72 s	0.06	2.92 s	2.89 s	∫ 5.16 s { 4.99 s
OCH,	3.75 s	0.09	3.96 s	3.98 s	3.94 s
3	3.72 s	0.42	3.86 s	3.88 s	$3.93 \ s$
OAc			2.45 s	2.47 s 2.16 s	_

J (Hz): 12,13 = 1;  $^{+}\Delta$ -values after addition of about 0.05 equivalents of Eu(fod)<sub>1</sub>.

genus Senecio. However, they are also present in several related genera. On the other hand there are several Senecio species which do not contain these compounds [1, 2]. Furthermore it is an open question whether such compounds are useful when considering the proposed new classification of the whole tribe [9]. The highly dehydrogenated cacalol derivatives, which are not widespread, but are typical of the genus Cacalia may be of special interest. Certainly further investigations, both from the chemical as well as from the botanical viewpoint, are necessary to arrive at any final conclusions.

## **EXPERIMENTAL**

IR spectra were measured in CCl<sub>4</sub> or CHCl<sub>3</sub>; <sup>1</sup>H NMR employed TMS as internal standard, assignments were established by extensive double resonance experiments; MS were determined at 70 eV; optical rotations were measured in CHCl<sub>3</sub>. The air dried plant material was extracted with Et<sub>2</sub>O-petrol (1:2) and the extracts were first separated by CC (Si gel, act. grade II) and further by repeated TLC (Si gel, GF 254) using Et<sub>2</sub>O-petrol mixtures as eluents. Known compounds were identified by comparison of their <sup>1</sup>H NMR and IR spectra with those of authentic material.

Senecio medley-woodii Hutch. (voucher 77/223). 125 g roots afforded 80 mg 1, 50 mg 2 (Et<sub>2</sub>O-petrol, 1:3) and 10 mg 4 (Et<sub>2</sub>O-petrol, 1:1), while 860 g aerial parts yielded 5 mg 6, 10 mg 1 and 3 mg 2.

Senecio halimifolius L. (voucher 77/345, Botanic Gardens Kirstenbosch). 470 g roots afforded 10 mg 8 (Et<sub>2</sub>O-petrol, 1:1) and 410 g aerial parts gave 200 mg 6, 200 mg 9, 5 mg 13, 300 mg 10 and 100 mg 11 (Et<sub>2</sub>O-petrol, 2:1).

Senecio inaequidens DC (voucher 77/6, 77/50, 77/66, 77/72, 77/77, 77/288). The different collections were extracted separately and the extracts were compared by TLC. No differences could be detected. 200 g roots afforded 10 mg 13, 60 mg 14, 30 mg 22 and 30 mg 23, while 500 g aerial parts yielded 50 mg 12, 20 mg 13, 600 mg 14, 600 mg 15 (Et<sub>2</sub>O-petrol, 1:3), 20 mg 16, 150 mg 23, 50 mg 24 (Et<sub>2</sub>O-petrol, 3:1 and  $CH_2Cl_2-Et_2O$ , 5:1) and ca 15 mg 25 (Et<sub>2</sub>O-petrol, 3:1).

Senecio inornatus DC (voucher 77/45). 950 g roots afforded 0.3 g 13, 2.8 g 18 (Et<sub>2</sub>O-petrol, 1:3), 5.6 g 17, 3.1 g 19 (Et<sub>2</sub>O-

petrol, 1:1) and 0.25 g 21, while 740 g aerial parts yielded 1 g 6, 1.1 g 18, 1 g 17, 1.2 g 19 and 0.5 g 20 (Et<sub>2</sub>O-petrol, 1:1).

 $3\beta$  - Acetoxy - 6β - (2 - methylbutyryloxy) - 10 βH - furanoere-mophilane (2). Colourless oil, IR  $\nu_{\rm max}$  cm<sup>-1</sup>: OAc 1740, 1250; CO<sub>2</sub>R 1730. MS m/e (rel. int.): 376.225 (M<sup>+</sup>, 5) (C<sub>22</sub>H<sub>32</sub>O<sub>5</sub>); 292 (M<sup>+</sup> - O=C=C (Me) Et, 24); 232 (292 - AcOH, 23); 85 (C<sub>4</sub>H<sub>9</sub>CO<sup>+</sup>, 47); 43 (H<sub>3</sub>C CO<sup>+</sup>, 100).

$$[\alpha]_{24^{\circ}}^{\frac{1}{2}} = \frac{589}{-47.3} \frac{578}{-49.6} \frac{546}{-57.0} \frac{436 \text{ nm}}{-103.4^{\circ}} (c = 4.31).$$

10 mg 2 in 2 ml MeOH were heated at  $60^\circ$  for 1 min with 0.5 ml 2N KOH. TLC afforded 5 mg 3, colourless oil. IR  $v_{max}$  cm<sup>-1</sup>: OH 3620; CO<sub>2</sub>R 1725.

3β - Acetoxy - 6β - (2 - methylbutyryloxy) - 10βH - furanoeremophil-9-one (4). Colourless oil, IR  $\nu_{\rm max}$  cm  $^{-1}$ : OAc 1745, 1245; CO  $_2$ R 1735; furan ketone 1685, 1535. MS m/e (rel. int.): 390.204 (M  $^+$ , 5) (C $_2$ H  $_{30}$ O  $_e$ ): 306 (M  $^+$  - O=C=C (Mc) Et, 32); 246 (306 - AcOH, 28); 85 (C $_4$ H $_9$ CO  $^+$ , 60); 43 (H $_3$ CCO  $^+$ , 100). 9 mg 4 were saponified as above. TLC afforded 5 mg 5, colourless oil, IR  $\nu_{\rm max}$  cm  $^{-1}$ : CO  $_2$ R 1740; furan ketone 1688, 1540. MS m/e (rel. int.): 348.194 (M  $^+$ , 4) (C $_{20}$ H $_{28}$ O $_{3}$ ); 330 (M  $^+$  - H $_2$ O, 4); 264 (M  $^+$  - O=C=C (Me) Et, 31); 246 (264 - H $_2$ O, 49); 85 (C $_4$ H $_9$ CO  $^+$ , 100).

6α - Angeloyloxy - 9, 10 - dehydrofuranoeremophil - 1 - one (8). Yellow coloured oil, UV  $\lambda_{max}$ : 346 nm (ether); IR  $\nu_{max}$  cm<sup>-1</sup>: C=CCO<sub>2</sub>R 1715, 1650; furan ketone 1670, 1610, 1570, 1525. MS m/e (rel. int.): 328.167 (M<sup>+</sup>, 44) (C<sub>20</sub>H<sub>24</sub>O<sub>4</sub>); 228 (M<sup>+</sup> - C<sub>4</sub>H<sub>7</sub>CO<sub>2</sub>H, 100); 83 (C<sub>4</sub>H<sub>7</sub>CO<sup>+</sup>, 81); 55 (83 - CO, 57).

12-Hydroxy-2, 3-dihydroeuparine (11). Colourless crystals, mp 77.5° (Et<sub>2</sub>O-petrol), UV  $\lambda_{\rm max}$  nm: 325, 273 ( $\epsilon$  = 12800, 20300) (ether); IR  $v_{\rm max}$  cm<sup>-1</sup>: OH 3620; hydrogen bonded ketone 3500–2600, 1640; NMR: 5.42 (dd(br), J = 9, 8, 2-H); 3.37 (dd(br), J = 16, 9, 3-H); 3.09 (ddd J = 16, 8, 1, 3'-H); 6.37 (s, 4-H); 7.50 (t, J = 1, 7-H); 2.54 (s, 9-H); 5.29 (brs, 11-H); 5.27 (brs 11'-H); 4.26 (AB q, 12-H), 12.98 (s, OH). MS m/e (rel int.): 234.089 (M<sup>+</sup>, 22) (C<sub>13</sub>H<sub>14</sub>O<sub>4</sub>); 219 (M<sup>+</sup> – Me, 8); 43 (C<sub>3</sub>H<sup>+</sup><sub>7</sub>, 100).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{-31.6} \frac{578}{-33.0} \frac{546}{-39.2} \frac{436 \text{ nm}}{-87.3^{\circ}} (c = 1.16).$$

14-Isovaleryloxy-O-methyl-1,2-dehydrocacalol (15). Colourless oil, IR  $\nu_{\text{max}}$  cm<sup>-1</sup> CO<sub>2</sub>R 1735; C=C 3060, 1610; furane 1570. MS m/e (rel. int.): 342.183 (M<sup>+</sup>, 46) (C<sub>21</sub>H<sub>26</sub>O<sub>4</sub>); 241 (M<sup>+</sup> - OCOR, 50); 240 (M<sup>+</sup> - RCO<sub>2</sub>H, 16); 225(240 - Me, 100); 85 (C<sub>4</sub>H<sub>9</sub>CO<sup>+</sup>, 5); 57 (85 - CO, 14).

$$[\alpha]_{24}^{\lambda} = \frac{589}{+51.0} \frac{578}{+54.0} \frac{546}{+66.0} \frac{436 \text{ nm}}{+174.0} (c = 1.2).$$

4-Hydroxy-3-methoxy-1-oxo-O-methyl-2,3-dehydrocacalol (24). Yellow crystals, mp 176° (CCl<sub>4</sub>), UV  $\lambda_{\text{max}}$  nm: 303, 225 ( $\epsilon$  = 9100, 11900) (ether); IR  $\nu_{\text{max}}$  cm<sup>-1</sup> OH 3590; C=C - C=O 1665, 1620; benzofurane 1588, 1542. MS m/e (rel. int.): 302.115 (M<sup>+</sup>, 37) (C<sub>17</sub>H<sub>18</sub>O<sub>5</sub>); 287 (M<sup>+</sup> - ·Me, 100); 284 (M<sup>+</sup> - H<sub>2</sub>O, 10); 273 (M<sup>+</sup> - ·CHO, 44); 272 (M<sup>+</sup> - CH<sub>2</sub>O, 36); 271 (M<sup>+</sup> - ·OMe, 89); 257 (272 - ·CH<sub>3</sub>, 26)

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589}{+7.0} \frac{578}{+7.2} \frac{546 \text{ nm}}{+8.1} (c = 0.73).$$

10 mg **24** were refluxed in 2 ml Ac<sub>2</sub>O for 1 hr. After TLC (CH<sub>2</sub>Cl<sub>2</sub>-Et<sub>2</sub>O, 5:1) 5 mg **28**, not free from **24**, were obtained, IR  $v_{\text{max}}$  cm<sup>-1</sup> C=C - C=O 1660, 1590; furane 1540. MS m/e (rel. int.): 284.105 (M<sup>+</sup>, 100) (C<sub>17</sub>H<sub>16</sub>O<sub>4</sub>); 269 (M<sup>+</sup> - ·Me, 28); 255 (M<sup>+</sup> - · CHO, 42).

14-Hydroxy-3-methoxy-1-oxo-O-methyl-2,3-dehydrocacalol (25). Yellow crystals, not free from 23, IR  $v_{\rm max}$  cm<sup>-1</sup>: OH 3600; C=C - C=O 1665, 1620. The mixture (8 mg) was heated for 1 hr in 2 ml Ac<sub>2</sub>O under reflux. TLC (CH<sub>2</sub>Cl<sub>2</sub>-Et<sub>2</sub>O, 5:1) afford 2 mg 26 and 2 mg 27 (Et<sub>2</sub>O-petrol 1:1).

Compound 26. Yellow crystals, mp 185° (ether), IR  $\nu_{\text{max}}$  cm<sup>-1</sup>: PhOAc 1775; benzofurane 1610, 1520. MS m/e (rel. int.): 328.131 (M<sup>+</sup>, 32) (C<sub>19</sub>H<sub>20</sub>O<sub>5</sub>), 286 (M<sup>+</sup> - O=C=CH<sub>2</sub>, 100); 271 (286 - CH<sub>3</sub>, 63).

Compound 27. Yellow crystals, mp 168° (ether), IR  $v_{\text{max}}$  cm<sup>-1</sup>: PhOAc 1775; OAc 1735, 1245; benzofurane 1610, 1520. MS m/e (rel. int.): 386.136 (M<sup>+</sup>, 70) (C<sub>21</sub>H<sub>22</sub>O<sub>7</sub>); 344 (M<sup>+</sup> - O=C = CH<sub>2</sub>, 100); 302 (344 - O=C=CH<sub>2</sub>, 32).

Calcalolpropionate (18). Colourless oil, IR  $\nu_{\text{max}}$  cm<sup>-1</sup>: PhOCOR 1765. MS m/e (rel. int.): 286.157 (M<sup>+</sup>) (C<sub>18</sub>H<sub>22</sub>O<sub>3</sub>); 230 (M<sup>+</sup> – O=C=CH Me, 100); 215 (230 – Me, 63).

14-Acetoxy-cacalolpropionate (19). Colourless oil, IR  $v_{\rm max}$  cm<sup>-1</sup> PhOCOR 1770; OAc 1735, 1230. MS m/e (rel. int.): 344.162 (M<sup>+</sup>) (C<sub>20</sub>H<sub>24</sub>O<sub>5</sub>); 302 (M<sup>+</sup> – O=C=CH<sub>2</sub>, 3); 288 (M<sup>+</sup> – O=C=CH Me, 70); 228 (288 – AcOH, 100).

14-Hydroxy-cacalolpropionate (20). Colourless oil, IR  $v_{\rm max}$  cm<sup>-1</sup>: OH 3600; PhOCOR 1770; benzofurane 1670, 1630, 1600. MS m/e (rel. int.): 302.152 (M<sup>+</sup>) ( $C_{18}H_{22}O_4$ ); 246 (M<sup>+</sup> – O=C=CH Me, 100); 228 (M<sup>+</sup> –  $C_2H_5CO_2H$ , 49); 231 (246 – Me, 25). 15 mg 20 were heated with 1 ml Ac<sub>2</sub>O for 30 min at 70°. After TLC 15 mg 19 were obtained, identical with the natural product.

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